

Common Atmospheric Pollutants + Sources

Mark Scheme 1

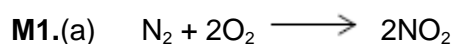
Level	GCSE (9-1)
Subject	Combined Science: Trilogy - Chemistry
Exam Board	AQA
Topic	5.9 Chemistry of the Atmosphere
Sub-Topic	Common Atmospheric Pollutants + Sources
Difficulty Level	Silver Level
Booklet	Mark Scheme 1

Time Allowed: 52 minutes

Score: /52

Percentage: /100

Grade Boundaries:



correct formulae for reactants

1

correct balancing

1

(b) $2.96 - 0.98$

correct values read from graph

1

$1.98 \div 2.96 (\times 100)$

allow ecf from readings from graph

1

$= 66.9(\%)$

allow 66.9 shown without working for the 3 calculation marks

1

incorrect number of sig. figs max 2 marks

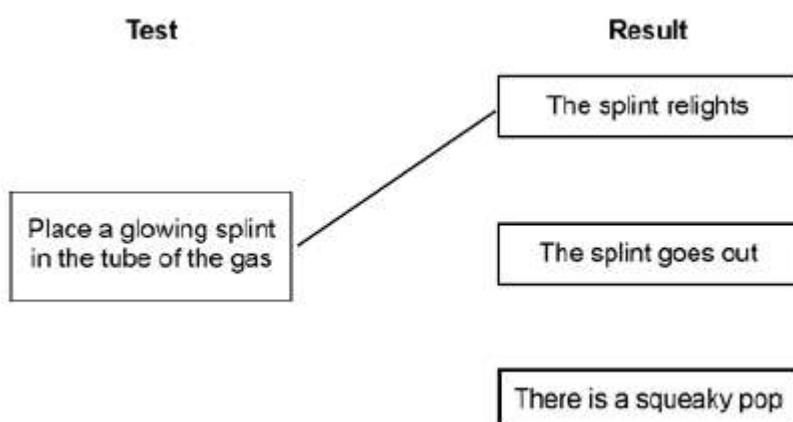
(c) less acid rain **or** fewer respiratory problems in humans

allow improved air quality

1

[6]

M2.(a)



more than one line from test negates the mark

1

(b) (i) place a lighted splint at the mouth of the tube 1

there is a squeaky pop
dependent on correct test 1

(ii) hydrogen is less reactive than magnesium
accept converse
accept magnesium is too reactive 1

(c) (i) any **one** from:

- to improve appearance or make it look nice
- to prevent corrosion
- to make it more durable
- cheaper than solid silver

 1

(ii) solution must be silver nitrate **or** contain silver ions 1

otherwise copper will be deposited **or** silver will not be deposited 1

spoon must be the negative electrode / cathode 1

because silver ions have a positive charge **or** go to negative electrode
or are discharged at the negative electrode. 1

(iii) because (plastic is an) insulator **or** does not conduct electricity
accept does not contain mobile electrons 1

[10]

M3.(a) (i) 2,4 drawn (as dots / crosses / e⁻)

1

(ii) Water (vapour) / steam
allow hydrogen oxide / H₂O
*do **not** accept hydroxide*

1

(b) any **two** pairs from:

carbon dioxide (1)

causes global warming (1)

*allow greenhouse effect / climate change / sea level rise /
melting of polar ice caps*

or

carbon (particles) / soot (1)

allow particulates

causes global dimming (1)

*allow blocks out sunlight / smog / prevents plant growth /
causes breathing difficulties*

or

carbon monoxide (1)

is toxic (1)

or

sulfur dioxide (1)

causes acid rain (1)

allow kills plants / erosion / acidifies water

4

[6]

M4.(a) (i) exothermic

*accept combustion
allow burning **or** oxidation **or**
redox*

1

- (ii) carbon monoxide / CO (is produced)
allow monoxide (is produced) ignore carbon oxide

1

because there is incomplete / partial combustion (of the fuel)
accept because there is insufficient oxygen / air (to burn the fuel)

1

- (b) Marks awarded for this answer will be determined by the Quality of Written Communication (QWC) as well as the standard of the scientific response. Examiners should also refer to the information in the [Marking guidance](#).

0 marksNo relevant content.

Level 1 (1-2 marks)There is a statement that crude oil is heated **or** that substances are cooled. However there is little detail and any description may be confused or inaccurate.

Level 2 (3-4 marks)There is some description of heating / evaporating crude oil **and either** fractions have different boiling points **or** there is an indication of a temperature difference in the column.

Level 3 (5-6 marks)There is a reasonable explanation of how petrol is or fractions are separated from crude oil using evaporating **and** condensing.

If cracking is given as a preliminary or subsequent process to fractional distillation then ignore.

However, if cracking / catalyst is given as part of the process, maximum is **level 2**.

Examples of chemistry points made in the response could include:

- Some / most of the hydrocarbons (or petrol) evaporate / form vapours
or gases

- When some of / a fraction of the hydrocarbons (or petrol) cool to their boiling point they condense
- Hydrocarbons (or petrol) that have (relatively) low boiling points and are collected near the top of the fractionating column or hydrocarbons with (relatively) high boiling points are collected near the bottom of the fractionating column
- The process is fractional distillation
- Heat the crude oil / mixture of hydrocarbons or crude oil / mixture is heated to about 350°C
- Some of the hydrocarbons remain as liquids
- Liquids flow to the bottom of the fractionating column
- Vapours / gases rise up the fractionating column
- Vapours / gases cool as they rise up the fractionating column
- The condensed fraction (or petrol) separates from the vapours / gases and flows out through a pipe
- Some of the hydrocarbons remain as vapours / gases
- Some vapours / gases rise out of the top of the fractionating column
- There is a temperature gradient in the fractionating column or the fractionating column is cool at the top and hot at the bottom

6

[9]

M5. (a) (i) C_7H_{16}
mark answer line first
answer may be given in the table

1

(ii) C_nH_{2n+2}

1

(b) (i) carbon monoxide
*do **not** accept carbon oxide*
*do **not** accept water*
ignore CO

1

- (ii) because of partial / incomplete combustion (in reaction 2) **or** complete combustion (in reaction 1)
*allow because there is less / insufficient oxygen (in reaction 2) **or** sufficient oxygen (in reaction 1) allow different amounts of oxygen used (in the reactions) **or** 19O_2 (in reaction 1) **and** 13O_2 (in reaction 2)*
ignore air
1
- (c) (i) 15 (%)
ignore units
1
- (ii) water (vapour)/steam
allow H_2O / OH_2 / hydrogen oxide
1
- (iii) sulfur in petrol / crude oil (reacts with oxygen)
it = sulfur dioxide
1
- (ii) because nitrogen **and** oxygen (are in the air and) react
*allow nitrogen **and** oxygen burn*
*accept nitrogen + oxygen \rightarrow nitrogen oxide **or** symbol equation*
ignore air
1
- at high temperature (inside a petrol engine)
allow heat / hot (engine)
1
- (d) because carbon dioxide / it causes global warming **or**
allow because carbon dioxide / it causes greenhouse effect / climate change
1

because carbon dioxide / it has an impact on oceans

because this carbon dioxide / carbon / it was 'locked up' (in fossil fuels) **or**

because the percentage/amount of carbon dioxide / it in the atmosphere is increasing

1

[11]

M6. (a) carbon dioxide decreased (by plants / trees)
allow plants / trees absorbed carbon dioxide

1

oxygen increased (by plants / trees)
allow plants / trees released oxygen
if neither of these marks awarded
allow plants / trees
photosynthesise for 1 mark

1

because coal 'locks up' / traps / stores carbon dioxide / carbon
allow trees 'locked up' carbon dioxide / carbon

1

(b) carbon / C

hydrogen / H

sulfur / S

all 3 correct 2 marks
1 or 2 correct 1 mark
allow H₂
ignore oxygen

2

(c) (i) 2 2

balancing must be correct

*do **not** accept changed formulae*

1

(ii) increases atmospheric pollution

carbon dioxide / CO₂ released

1

from the (thermal) decomposition of calcium carbonate **or**

*accept causes global warming **or** CO₂ is a greenhouse gas*

description of this decomposition **or** equation

ignore sulfur dioxide and effects in this part

1

decreases atmospheric pollution

sulfur dioxide / SO₂ is removed

accept less acid rain produced

1

by reaction with calcium oxide **or** calcium carbonate

*accept neutralisation **or** forms calcium sulfate*

1

[10]